

Department of Materials Science and Engineering

CORROSION AND MATERIALS PROTECTION LABORATORY

Analytical and non-analytical instruments



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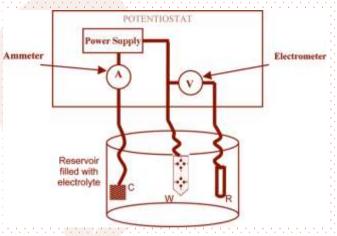
INSTRUMENTS

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Potentiostat/Galvanostat

Introduction

In order for an electrochemical process to take place, there must be an anode, a cathode, as well as both an ionic and electrical conduction path between the two. When performing a DC polarization scan, as will be discussed in this application note, the ionic conduction path is provided through the solution separating the working and counter electrodes, while the



electrical conduction path is provided through *Schematic presentation of a electrochemical test cell* the **potentiostat**. This potentiostat is then used to control the driving force for electrochemical reactions taking place on the working electrode. The magnitude of this driving force in turn dictates which electrochemical processes actually take place at the anode and cathode, as well as their rate. The potential, is a thermodynamic property and provides information on whether or not a reaction can occur. The current, on the other hand, represents the rate with which the anodic or cathodic reactions are taking place on the working electrode. Typically, the current density is used rather than current itself.

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Potentiostat/Galvanostat

Introduction

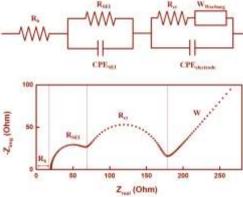


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Electrochemical corrosion experiments measure and/or control the potential and current of the oxidation/reduction reactions. Several types of experiments are possible by manipulating and measuring these two variables.

Most experiments impose a potential on the working electrode and measure the resulting current, such as: potentiostatic and potentiodynamic experiments. Experiments where the current is imposed rather than the potential are referred to as **galvanodynamic** or galvanostatic.



Equivalent corrosion circuit and corresponding Nyquist diagram

Electrochemical impedance spectroscopy (EIS) is a powerful technique that utilizes a small amplitude, alternating current (AC) signal to probe the impedance characteristics of a cell. The AC signal is scanned over a wide range of frequencies to generate an impedance spectrum for the electrochemical cell under test. EIS differs from direct current (DC) techniques in that it allows the study of capacitive, inductive, and diffusion processes taking place in the electrochemical cell.

Potentiostat/Galvanostat

Introduction



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As mentioned earlier, potentiostats are used to keep the potential (voltage) between a working electrode

and a reference electrode at a constant value. Galvanostats are used to maintain a constant flow of

current through an electrolytic cell.

Potentiostats and galvanostats are electrochemical instruments used in electrochemistry, battery and fuel

cell testing, corrosion control, voltammetry, biomedical research, surface imaging, and related

applications.



Device: Potentiostat/Galvanostat

Manufacturer: METROHMAG



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This lab is equipped with two potentiostat/galvanostat devices which the technical specification for each will be presented.

Technical specification

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Electrode connections	2, 3
Potential range	± 5 V
Compliance voltage	± 12 V
Maximum current	± 80 mA
Current ranges	10 mA to 10 nA
Potential accuracy	± 0.2 %
Potential resolution	3 µV
Current accuracy	± 0.2 %
Current resolution	0.0003 % (of curr
Input impedance	> 100 GOhm
Potentiostat bandwidth	500 kHz
Computer interface	USB
Control software	NOVA
impedance measurements frequency	10 µHz - 500 kHz



Device: Potentiostat/Galvanostat

Manufacturer: IVIUM TECHNOLOGIES

Technical specification

Current compliance	±5A
Maximum compliance voltage	±10V
Maximum applied voltage	±10V below 1A, and ±8V up
	to 5A
Electrode connections	4; WE, CE, RE, S (and GND)
	with 4mm banana plugs
Potentiostat bandwidth	8 MHz
Stability settings	High Speed, Standard and High
	Stability
Programmable response filter	1 MHz, 100 kHz, 10 kHz, 1
	kHz, 10 Hz
Signal acquisition	Dual channel 24bit ADC, 100,000
	samples/s
Applied potential range	±10 V, 0.02 mV res. (20bit)
Applied potential accuracy	0,2% or 1 mV
Current ranges	±1 pA to ±10 A



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Device: Potentiostat/Galvanostat

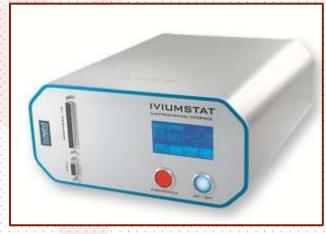
Manufacturer: IVIUM TECHNOLOGIES

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Technical specification (cont.)

Measured current resolution	0.00001% of current range, minimum 0.6 aA
Measured current accuracy	10 pA + 0.025% of FSR
Applied current resolution	0.00013% of applied
	current range
Applied current accuracy	0.2%
Galvanostatic current ranges	±100 pA to ±10 A
Potential ranges	±1 mV, ±4 mV, ±10 mV,
	±40 mV, ±100 mV, ±400
	mV, ±1 V, ±4 V, ±10 V
Measured potential resolution	0.00001% of potential
	range, minimum 0.15 nV
Measured potential accuracy	0.2% or 1 mV
Frequency range	10 µHz to 8 MHz
Amplitude	0.015 mV to 1.0 V, or
	0.03% to 100% of
	current range



Technical specification (cont.)

Input impedance	i.	ł.	1	į,	ġ	ł
Input bias current	ł	i.	j	ł		ł
Bandwidth	j.	ł.	ł	i)	ġ	
Ohmic drop comp	en	sa	tic	m		ł
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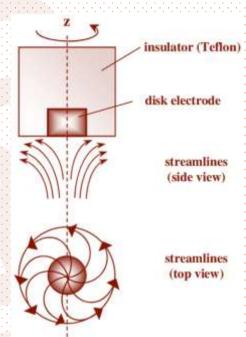
Dynamic range

>1000Gohm //<8pF	
<10pA	
>16MHz	
2V/current range, 16	bit
resolution	
0.05nV to 10V, and	
0.2aA to 5A	

Rotating disc electrode

Introduction

The rotating disc electrode (RDE) is the classical hydrodynamic electroanalytical technique used to limit the diffusion layer thickness. Forced convection has several advantages which include: 1) the rapid establishment of a high rate of steady-state mass transport and 2) easily and reproducibly controlled convection over a wide range of mass transfer coefficients. The RDE consists of a disc (e.g. of Pt, Ni, Cu, Au, Fe, Si, CdS, GaAs, glassy carbon and graphite) set into an insulating (PTFE) surround. The electrode is rotated about its vertical axis, typically between 400 and 10,000 rpm. The theory for the hydrodynamics at the RDE assumes that the electrode is uniformly accessible



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Schematic representation of streamlines during RDE experiment

and affords a precise and reproducible control of the convection and diffusion of reactant to the electrode. Hence, the RDE can be used to study the kinetics of interfacial processes. This lab is equipped with a RDE which its specifications are introduced.

Device: Rotating disc electrode

Manufacturer: METROHMAG

This lab is equipped with a rotating disc electrode device with the following specifications.

Technical specification (cont.)

Rotation rate	100-10000 rpm
Rotation rate precision	1 rpm
Acceleration and deceleration rate	4000 rpm/s





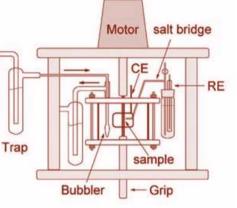
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Slow strain rate testing

Introduction

Slow strain rate testing (SSRT)), also called constant extension rate tensile testing (CERT), is a popular test used by research scientists to study stress corrosion cracking. It involves a slow (compared to conventional tensile tests) dynamic strain applied at a constant extension rate in the environment of interest. These test results are compared to those for similar tests in a, known to be inert, environment. The evaluated parameters are:

- time to specimen failure (e.g., breakage, or from other where the specimen is held in a corrosion test cell "failure" criteria)
- ductility (by elongation to fracture or the reduction of the area)
- ultimate tensile strength (from the maximum load)
- area under the elongation load curve (which represents the fracture energy)
- percent of ductile/brittle fracture on the fracture surface
- threshold stress for cracking



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Schematic representation of SSRT test

Slow strain rate testing

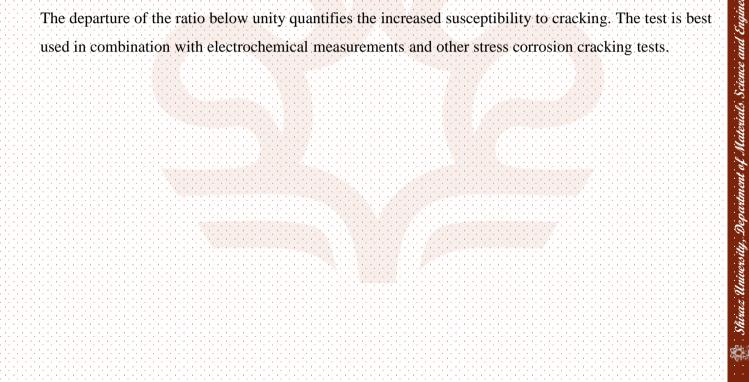
Introduction

The results of the SSRT tests are evaluated using the ratio:

result from specimen in test enviroment result from specimen in inert enviroment

The departure of the ratio below unity quantifies the increased susceptibility to cracking. The test is best

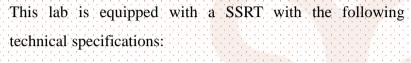
used in combination with electrochemical measurements and other stress corrosion cracking tests.





Device: Slow strain rate testing machine

Manufacturer: ROYIN GRAN SANAT GHARB ASIA



Technical specification

Maximum load	5 ton
Load precision	1 kg
Strain rate	10-7-10-4 (1/s)
Strain rate precision	1%
Computer interface	USB
Output format	txt, xlsx, html
• Continues monitoring of load an	d displacement and plotting

Continues monitoring of load and displacement and

force-displacement curve



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